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------ISO 9001: 2015------

MATERIAL SAFETY DATA SHEET

1.Identification 1.1GHS Product identifier Ferulic acid, 98% Code F 1405

2.Hazard identification

2.1Classification of the substance or mixture

Skin irritation, Category 2 Eye irritation, Category 2

Specific target organ toxicity \u2013 single exposure, Category 3

2.2GHS label elements, including precautionary statements

Pictogram(s)



Signal word

Warning

Hazard statement(s)

H315 Causes skin irritation

H319 Causes serious eye irritation

H335 May cause respiratory irritation

Precautionary statement(s)

Prevention

P264 Wash ... thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye

protection/face protection.

P261 Avoid breathing dust/fume/gas/mist/vapours/spray. P271 Use only outdoors or in a well-ventilated area. P302+P352 IF ON SKIN: Wash with plenty of water/...

Response P302+P352 IF ON SKIN: Wash with plenty of P321 Specific treatment (see ... on this label).

P332+P313 If skin irritation occurs: Get medical advice/attention. P362+P364 Take off contaminated clothing and wash it before

reuse.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to

do. Continue rinsing.

P337+P313 If eye irritation persists: Get medical advice/attention. P304+P340 IF INHALED: Remove person to fresh air and keep

comfortable for breathing.

P312 Call a POISON CENTER/doctor/\u2026if you feel unwell.
P403+P233 Store in a well-ventilated place. Keep container tightly

P403+P closed.

P405 Store locked up.

Disposal P501 Dispose of contents/container to ...

2.30ther hazards which do not result in classification

none

Storage

3. Composition/information on ingredients

3.1Substances

C. (Cabotalioco					
Chemical name	Common names and synonyms	CAS number	EC number	Concentration	
Trans-Ferulic Acid	Trans-Ferulic Acid	537-98-4	none	100%	

4.First-aid measures

4.1Description of necessary first-aid measures

General advice

Consult a physician. Show this safety data sheet to the doctor in attendance.

If inhaled

If breathed in, move person into fresh air. If not breathing, give artificial respiration. Consult a physician.

In case of skin contact

Wash off with soap and plenty of water. Consult a physician.

In case of eye contact

Rinse thoroughly with plenty of water for at least 15 minutes and consult a physician.

If swallowed

Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

4.2Most important symptoms/effects, acute and delayed

no data available

4.3Indication of immediate medical attention and special treatment needed, if necessary

/SRP:/ Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on the left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. /Poisons A and B/

5. Fire-fighting measures

5.1Extinguishing media

Suitable extinguishing media

Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

5.2Specific hazards arising from the chemical

no data available

5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

6.Accidental release measures

6.1Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Avoid dust formation. Avoid breathing vapours, mist or gas. Ensure adequate ventilation.

Evacuate personnel to safe areas. Avoid breathing dust. For personal protection see section 8.

6.2Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

6.3Methods and materials for containment and cleaning up

Pick up and arrange disposal. Sweep up and shovel. Keep in suitable, closed containers for disposal.

7. Handling and storage

7.1Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Avoid exposure - obtain special instructions before use. Provide appropriate exhaust ventilation at places where dust is formed. For precautions see section 2.2.

7.2Conditions for safe storage, including any incompatibilities

Store in cool place. Keep container tightly closed in a dry and well-ventilated place.

8. Exposure controls/personal protection

8.1Control parameters

Occupational Exposure limit values

no data available

Biological limit values

no data available

8.2Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday. 8.3Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Safety glasses with side-shields conforming to EN166. Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

Skin protection

Wear impervious clothing. The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace. Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique(without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Respiratory protection

Wear dust mask when handling large quantities.

Thermal hazards

no data available

9. Physical and chemical properties

Physical state slightly yellow powder Colour no data available Odour no data available Melting point/ freezing point 168-172\u00baC

Boiling point or initial boiling 372.3\u00baC at 760 mmHg

point and boiling range

Flammability no data available Lower and upper explosion no data available

limit / flammability limit

Flash point 150.5\u00baC
Auto-ignition temperature
Decomposition temperature
pH no data available
no data available
Kinematic viscosity no data available

Solubility In water, 5.97X10+3 mg/L at 25\u00b0C (est)

Partition coefficient n- log Kow = 1.51

octanol/water (log value)

Vapour pressure 2.69X10-6 mm Hg at 25\u00b0C (est)

Density and/or relative 1.316 g/cm3

density

Relative vapour density no data available Particle characteristics no data available

10.Stability and reactivity

10.1Reactivity
no data available
10.2Chemical stability

Stable under recommended storage conditions.

10.3Possibility of hazardous reactions

no data available

10.4Conditions to avoid no data available

10.5Incompatible materials

no data available

10.6Hazardous decomposition products

no data available

11.Toxicological information

Acute toxicity
Oral: no data available
Inhalation: no data available
Dermal: no data available
Skin corrosion/irritation
no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available
Germ cell mutagenicity
no data available
Carcinogenicity
no data available
Reproductive toxicity
no data available
STOT-single exposure
no data available
STOT-repeated exposure
no data available
Aspiration hazard

12. Ecological information

12.1Toxicity

no data available

Toxicity to fish: no data available

Toxicity to daphnia and other aquatic invertebrates: no data available

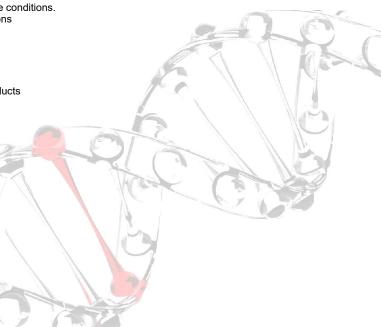
Toxicity to algae: no data available

Toxicity to microorganisms: no data available

12.2Persistence and degradability

AEROBIC: Ferulic acid, present at 100 ppm in Greenfield (California) sandy loam with a pH of 7, underwent 60-72% decomposition via 14-CO2 evolution in 28 days(1). In a similar study, ferulic acid, present at 100 ppm in Chino (California) loam with a pH of 5.6, underwent 77% decomposition via 14-CO2 evolution after 28 days(2); in San Jacinto (California) sandy loam with a pH of 8, 100 ppm ferulic acid underwent 13% decomposition by 14-CO2 evolution over 28 days(2). This lower level of decomposition was thought to be due to polymerization of ferulic acid to humic acid type compounds(2). Therefore, under neutral and acidic conditions in soil, ferulic acid is expected to biodegrade rapidly; under alkaline conditions in soil, biodegradation of ferulic acid is not expected to be rapid.

12.3Bioaccumulative potential



An estimated BCF of 3.2 was calculated in fish for ferulic acid, using a measured log Kow of 1.51 (1) and a regression-derived equation (2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organism is low(SRC), provided the compound is not metabolized by the organism(SRC).

12.4Mobility in soil

The Koc of ferulic acid is estimated as 57(SRC), using a log Kow of 1.51(1)(SRC) and a regression-derived equation(2). According to a classification scheme(3), this estimated Koc value suggests that ferulic acid is expected to have high mobility in soil. The pKa of ferulic acid is 4.58(4), indicating that this compound will almost entirely exist in the anion form in the environment. Anions generally do not adsorb more strongly to soils containing organic carbon and clay, in comparison with their neutral counterparts(5). 12.50ther adverse effects no data available

13.Disposal considerations

13 1Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems. Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

14. Transport information

14.1UN Number

ADR/RID: no data available

14.2UN Proper Shipping Name
ADR/RID: no data available

IMDG: no data available

IMDG: no data available

IMDG: no data available

IATA: no data available

IATA: no data available

IATA: no data available

IATA: no data available

IMDG: no data available

IATA: no data available

14.5Environmental hazards

ADR/RID: no IMDG: no IATA: no

14.6Special precautions for user

no data available

14.7Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code no data available

15.Regulatory information

15.1Safety, health and environmental regulations specific for the product in guestion

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Chemical name	Common names and synonyms	CAS number	EC number
Trans-Ferulic Acid	Trans-Ferulic Acid	537-98-4	none
European Inventory of	Listed.		
EC Inventory	Listed.		
United States Toxic Su	Not Listed.		
China Catalog of Haza	Not Listed.		
New Zealand Inventory	Not Listed.		
Philippines Inventory of	Not Listed.		
Vietnam National Chen	Not Listed.		
Chinese Chemical Inve	Not Listed.		

Section 16: Other Information

This safety data sheet should be used in conjunction with technical sheets. It does not replace them. The information given is based on our knowledge of this product, at the time of publication. It is given in good faith. The attention of the user is drawn to the possible risks incurred by using the product for any other purpose other than that for which it was intended. This does not in any way excuse the user from knowing and applying all the regulations governing his activity. It is the sole responsibility of the user to take all precautions required in handling the product. The aim of the mandatory regulations mentioned is to help the user to fulfill his obligations regarding the use of hazardous products.