

# OTTO CHEMIE PVT LTD

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ISO 9001: 2015

## MATERIAL SAFETY DATA SHEET

### 1. Identification

1.1 GHS Product identifier  
2-Dimethylaminoethanol, 98%  
Code D 1975

### 2. Hazard identification

2.1 Classification of the substance or mixture  
Flammable liquids, Category 3  
Acute toxicity - Oral, Category 4  
Acute toxicity - Dermal, Category 4  
Skin corrosion, Category 1B  
Acute toxicity - Inhalation, Category 4  
2.2 GHS label elements, including precautionary statements  
Pictogram(s)



Signal word

Danger

Hazard statement(s)

H226 Flammable liquid and vapour  
H302 Harmful if swallowed  
H312 Harmful in contact with skin  
H314 Causes severe skin burns and eye damage  
H332 Harmful if inhaled

Precautionary statement(s)

Prevention

P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.  
P233 Keep container tightly closed.  
P240 Ground and bond container and receiving equipment.  
P241 Use explosion-proof [electrical/ventilating/lighting/...] equipment.  
P242 Use non-sparking tools.  
P243 Take action to prevent static discharges.  
P280 Wear protective gloves/protective clothing/eye protection/face protection.  
P264 Wash ... thoroughly after handling.  
P270 Do not eat, drink or smoke when using this product.  
P260 Do not breathe dust/fume/gas/mist/vapours/spray.  
P261 Avoid breathing dust/fume/gas/mist/vapours/spray.  
P271 Use only outdoors or in a well-ventilated area.  
P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].  
P370+P378 In case of fire: Use ... to extinguish.  
P301+P312 IF SWALLOWED: Call a POISON CENTER/doctor if you feel unwell.  
P330 Rinse mouth.  
P302+P352 IF ON SKIN: Wash with plenty of water/...

Response

P312 Call a POISON CENTER/doctor if you feel unwell.  
P321 Specific treatment (see ... on this label).  
P362+P364 Take off contaminated clothing and wash it before reuse.  
P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.  
P363 Wash contaminated clothing before reuse.  
P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.  
P310 Immediately call a POISON CENTER/doctor if you feel unwell.  
P305+P351+P338 IF IN EYES: Rinse cautiously with water for

Storage several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.  
P403+P235 Store in a well-ventilated place. Keep cool.  
P405 Store locked up.  
Disposal P501 Dispose of contents/container to ...  
2.3 Other hazards which do not result in classification  
none

### 3. Composition/information on ingredients

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
N,N-dimethylethanolamine	N,N-dimethylethanolamine	108-01-0	none	100%

#### 4. First-aid measures

##### 4.1 Description of necessary first-aid measures

###### General advice

Consult a physician. Show this safety data sheet to the doctor in attendance.

###### If inhaled

Fresh air, rest. Half-upright position. Refer for medical attention.

###### In case of skin contact

First rinse with plenty of water for at least 15 minutes, then remove contaminated clothes and rinse again. Refer for medical attention .

###### In case of eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

###### If swallowed

Rinse mouth. Do NOT induce vomiting. Give one or two glasses of water to drink. Rest. Refer for medical attention .

##### 4.2 Most important symptoms/effects, acute and delayed

Inhalation of the vapor or mist can cause irritation to the upper respiratory tract. Asthmatic symptoms have been reported. Extremely irritating; may cause permanent eye injury. Corrosive; will cause severe skin damage with burns and blistering. Ingestion may cause damage to the mucous membranes and gastrointestinal tract. (USCG, 1999)

##### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on the left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. /Organic bases/Amines and related compounds/

#### 5. Fire-fighting measures

##### 5.1 Extinguishing media

Suitable extinguishing media

Water may be ineffective. Alcohol foam.

##### 5.2 Specific hazards arising from the chemical

Special Hazards of Combustion Products: May contain toxic gases including ammonia (incomplete combustion) and NOx. Behavior in Fire: Produces gaseous nitrogen compounds that are highly toxic and irritating. (USCG, 1999)

##### 5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

#### 6. Accidental release measures

##### 6.1 Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Avoid dust formation. Avoid breathing vapours, mist or gas. Ensure adequate ventilation.

Evacuate personnel to safe areas. Avoid breathing dust. For personal protection see section 8.

##### 6.2 Environmental precautions

Personal protection: gas-tight chemical protection suit including self-contained breathing apparatus. Collect leaking and spilled liquid in sealable non-metallic containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

##### 6.3 Methods and materials for containment and cleaning up

Contain spillage, and then collect with an electrically protected vacuum cleaner or by wet-brushing and place in container for disposal according to local regulations.

#### 7. Handling and storage

##### 7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Avoid exposure - obtain special instructions before use. Provide appropriate exhaust ventilation at places where dust is formed. For precautions see section 2.2.

##### 7.2 Conditions for safe storage, including any incompatibilities

Fireproof. Separated from strong oxidants, acids, acid chlorides, copper and food and feedstuffs. Do not store near acids.

#### 8. Exposure controls/personal protection

##### 8.1 Control parameters

## Occupational Exposure limit values

no data available

## Biological limit values

no data available

## 8.2 Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

## 8.3 Individual protection measures, such as personal protective equipment (PPE)

### Eye/face protection

Safety glasses with side-shields conforming to EN166. Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

### Skin protection

Wear impervious clothing. The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace. Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

### Respiratory protection

Wear dust mask when handling large quantities.

### Thermal hazards

no data available

## 9. Physical and chemical properties

Physical state clear to pale yellow liquid

Colour Colorless liquid

Odour Amine odor

Melting point/ freezing point  $-70^{\circ}\text{C}$

Boiling point or initial boiling point and boiling range  $139^{\circ}\text{C}$

Flammability Flammable. Gives off irritating or toxic fumes (or gases) in a fire.

Lower and upper explosion limit / flammability limit Lower 1.6% Upper 11.9%

Flash point  $40^{\circ}\text{C}$

Auto-ignition temperature  $295^{\circ}\text{C}$  (USCG, 1999)

Decomposition temperature no data available

pH no data available

Kinematic viscosity 3.5839 mPa.s at  $21.6^{\circ}\text{C}$

Solubility In water: miscible

Partition coefficient n-octanol/water (log value)  $\log K_{ow} = -0.55$  at  $23^{\circ}\text{C}$

Vapour pressure 100 mm Hg ( $55^{\circ}\text{C}$ )

Density and/or relative density 0.888

Relative vapour density 3.03 (Relative to Air)

Particle characteristics no data available

## 10. Stability and reactivity

### 10.1 Reactivity

no data available

### 10.2 Chemical stability

Stable under recommended storage conditions.

### 10.3 Possibility of hazardous reactions

Flammable liquid when exposed to heat or flame; can react vigorously with oxidizing materials. Ignites spontaneously in contact with cellulose nitrate of high surface area. The vapour is heavier than air. DIMETHYLAMINOETHANOL is an organic compound with both amine and alcohol substituents. Amines are chemical bases. They neutralize acids to form salts plus water. These acid-base reactions are exothermic. The amount of heat that is evolved per mole of amine in a neutralization is largely independent of the strength of the amine as a base. Amines may be incompatible with isocyanates, halogenated organics, peroxides, phenols (acidic), epoxides, anhydrides, and acid halides. Flammable gaseous hydrogen is generated by amines in combination with strong reducing agents, such as hydrides. This compound may react vigorously with oxidizing materials.

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Oxidizing agents, Copper, Zinc, Iron, Do not store near acids.

### 10.6 Hazardous decomposition products

When heated to decomposition it emits toxic fumes of  $\text{NO}_x$ .

## 11. Toxicological information

### Acute toxicity

Oral: LD50 rat oral 2000 mg/kg

Inhalation: LC50 Rats inhalation 1641 (862-3125) ppm/4 hr

Dermal: no data available

Skin corrosion/irritation  
no data available  
Serious eye damage/irritation  
no data available  
Respiratory or skin sensitization  
no data available  
Germ cell mutagenicity  
no data available  
Carcinogenicity  
no data available  
Reproductive toxicity  
no data available  
STOT-single exposure  
no data available  
STOT-repeated exposure  
no data available  
Aspiration hazard  
no data available

## 12. Ecological information

### 12.1 Toxicity

Toxicity to fish: no data available  
Toxicity to daphnia and other aquatic invertebrates: no data available  
Toxicity to algae: no data available  
Toxicity to microorganisms: no data available

### 12.2 Persistence and degradability

AEROBIC: 2-Dimethylaminoethanol, present at 100 mg/L, reached 60.5% of its theoretical BOD in 2 weeks using an activated sludge inoculum at 30 mg/L in the Japanese MITI test which classified the compound as readily biodegradable(1). A static test using a non adapted activated sludge inoculum measured a 2-dimethylaminoethanol degradation of >90% in 13 days(2).

### 12.3 Bioaccumulative potential

An estimated BCF of 3 was calculated in fish for 2-dimethylaminoethanol(SRC), using a log Kow of -0.55(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is low(SRC).

### 12.4 Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc of 2-dimethylaminoethanol can be estimated to be 1(SRC). According to a classification scheme(2), this estimated Koc value suggests that 2-dimethylaminoethanol is expected to have very high mobility in soil. The pKa of 2-dimethylaminoethanol is 9.3(3). At pH 8.2, 20% will be the free amine. Thus, this compound will exist almost entirely in the cation form in the environment and cations generally adsorb to organic carbon and clay more strongly than their neutral counterparts(4). As a result, 2-dimethylaminoethanol may have greater adsorption and less mobility than its estimated Koc value indicates.

### 12.5 Other adverse effects

no data available

## 13. Disposal considerations

### 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

#### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## 14. Transport information

### 14.1 UN Number

ADR/RID: UN2051                      IMDG: UN2051                      IATA: UN2051

### 14.2 UN Proper Shipping Name

ADR/RID: 2-DIMETHYLAMINOETHANOL

IMDG: 2-DIMETHYLAMINOETHANOL

IATA: 2-DIMETHYLAMINOETHANOL

### 14.3 Transport hazard class(es)

ADR/RID: 8                                      IMDG: 8                                      IATA: 8

### 14.4 Packing group, if applicable

ADR/RID: II                                      IMDG: II                                      IATA: II

### 14.5 Environmental hazards

ADR/RID: no                                      IMDG: no                                      IATA: no

### 14.6 Special precautions for user

no data available

### 14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

no data available

## 15. Regulatory information

### 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
N,N-dimethylethanolamine	N,N-dimethylethanolamine	108-01-0	none
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.

### Section 16: Other Information

This safety data sheet should be used in conjunction with technical sheets. It does not replace them. The information given is based on our knowledge of this product, at the time of publication. It is given in good faith. The attention of the user is drawn to the possible risks incurred by using the product for any other purpose other than that for which it was intended. This does not in any way excuse the user from knowing and applying all the regulations governing his activity. It is the sole responsibility of the user to take all precautions required in handling the product. The aim of the mandatory regulations mentioned is to help the user to fulfill his obligations regarding the use of hazardous products.

