

# OTTO CHEMIE PVT LTD

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ISO 9001: 2015

## MATERIAL SAFETY DATA SHEET

### 1. Identification

#### 1.1 GHS Product identifier

Dekalin, puriss, 99%+

Code: D 1267

### 2. Hazard identification

#### 2.1 Classification of the substance or mixture

Flammable liquids, Category 3

Aspiration hazard, Category 1

Skin corrosion, Category 1C

Acute toxicity - Inhalation, Category 3

Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 2

#### 2.2 GHS label elements, including precautionary statements

#### Pictogram(s)



#### Signal word

#### Hazard statement(s)

Danger

H226 Flammable liquid and vapour

H304 May be fatal if swallowed and enters airways

H314 Causes severe skin burns and eye damage

H331 Toxic if inhaled

H411 Toxic to aquatic life with long lasting effects

#### Precautionary statement(s)

#### Prevention

P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.

P233 Keep container tightly closed.

P240 Ground and bond container and receiving equipment.

P241 Use explosion-proof [electrical/ventilating/lighting/...] equipment.

P242 Use non-sparking tools.

P243 Take action to prevent static discharges.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

P264 Wash ... thoroughly after handling.

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P271 Use only outdoors or in a well-ventilated area.

P273 Avoid release to the environment.

#### Response

P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].

P370+P378 In case of fire: Use ... to extinguish.

P301+P310 IF SWALLOWED: Immediately call a POISON CENTER/doctor/2026

P331 Do NOT induce vomiting.

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P363 Wash contaminated clothing before reuse.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P310 Immediately call a POISON CENTER/doctor/2026

P321 Specific treatment (see ... on this label).

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P311 Call a POISON CENTER/doctor/2026

P391 Collect spillage.

#### Storage

P403+P235 Store in a well-ventilated place. Keep cool.

P405 Store locked up.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

#### Disposal

P501 Dispose of contents/container to ...

#### 2.3 Other hazards which do not result in classification

none

### 3. Composition/information on ingredients

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
decalin	decalin	91-17-8	none	100%

#### 4. First-aid measures

##### 4.1 Description of necessary first-aid measures

###### General advice

Consult a physician. Show this safety data sheet to the doctor in attendance.

###### If inhaled

Fresh air, rest. Refer for medical attention.

###### In case of skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention .

###### In case of eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

###### If swallowed

Rinse mouth. Do NOT induce vomiting. Give one or two glasses of water to drink. Refer for medical attention .

##### 4.2 Most important symptoms/effects, acute and delayed

Inhalation or ingestion irritates nose and throat, causes numbness, headache, vomiting; urine may become blue. Irritates eyes.

Liquid de-fats skin and causes cracking and secondary infection; eczema may develop. (USCG, 1999)

##### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

Basic treatment: Establish a patent airway. Suction if necessary. Watch for signs of respiratory insufficiency and assist ventilations if necessary. Administer oxygen by nonrebreather mask at 10 to 15 L/min. Monitor for shock and treat if necessary ... . Anticipate seizures and treat if necessary ... . For eye contamination, flush eyes immediately with water. Irrigate each eye continuously with normal saline during transport ... . Do not use emetics. For ingestion, rinse mouth and administer 5 ml/kg up to 200 ml of water for dilution if the patient can swallow, has a strong gag reflex, and does not drool. Administer activated charcoal ... . /Naphthalene and Related Compounds/

#### 5. Fire-fighting measures

##### 5.1 Extinguishing media

###### Suitable extinguishing media

TO FIGHT FIRE USE FOAM, CARBON DIOXIDE, DRY CHEMICAL.

##### 5.2 Specific hazards arising from the chemical

Excerpt from ERG Guide 130 [Flammable Liquids (Water-Immiscible / Noxious)]: HIGHLY FLAMMABLE: Will be easily ignited by heat, sparks or flames. Vapors may form explosive mixtures with air. Vapors may travel to source of ignition and flash back. Most vapors are heavier than air. They will spread along ground and collect in low or confined areas (sewers, basements, tanks). Vapor explosion hazard indoors, outdoors or in sewers. Those substances designated with a (P) may polymerize explosively when heated or involved in a fire. Runoff to sewer may create fire or explosion hazard. Containers may explode when heated. Many liquids are lighter than water. (ERG, 2016)

##### 5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

#### 6. Accidental release measures

##### 6.1 Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Avoid dust formation. Avoid breathing vapours, mist or gas. Ensure adequate ventilation.

Evacuate personnel to safe areas. Avoid breathing dust. For personal protection see section 8.

##### 6.2 Environmental precautions

Personal protection: chemical protection suit including self-contained breathing apparatus. Do NOT let this chemical enter the environment. Ventilation. Collect leaking liquid in sealable containers. Carefully collect remainder. Then store and dispose of according to local regulations.

##### 6.3 Methods and materials for containment and cleaning up

Pick up and arrange disposal. Sweep up and shovel. Keep in suitable, closed containers for disposal.

#### 7. Handling and storage

##### 7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Avoid exposure - obtain special instructions before use. Provide appropriate exhaust ventilation at places where dust is formed. For precautions see section 2.2.

##### 7.2 Conditions for safe storage, including any incompatibilities

Fireproof. Separated from oxidants. Cool. Keep in the dark. Well closed. Handle and store under Nitrogen. ...Potentially explosive peroxides can form on long time storage in contact with air. Light and heat accelerate peroxide /formation/.

#### 8. Exposure controls/personal protection

##### 8.1 Control parameters

###### Occupational Exposure limit values

no data available

###### Biological limit values

no data available

##### 8.2 Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

##### 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Safety glasses with side-shields conforming to EN166. Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

#### Skin protection

Wear impervious clothing. The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace. Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

#### Respiratory protection

Wear dust mask when handling large quantities.

#### Thermal hazards

no data available

### 9. Physical and chemical properties

Physical state	clear, colorless liquid
Colour	Clear colorless liquid
Odour	Slight odor resembling menthol; pure decalin does not smell of naphthalene
Melting point/ freezing point	-31\u00baC
Boiling point or initial boiling point and boiling range	189-191\u00baC(lit.)
Flammability	Flammable.
Lower and upper explosion limit / flammability limit	LOWER 0.7% @ 100 DEG C; UPPER 4.9% @ 100 DEG C
Flash point	57\u00baC
Auto-ignition temperature	250\u00baC
Decomposition temperature	no data available
pH	no data available
Kinematic viscosity	1.788 cP @ 70 deg F
Solubility	In water:6 mg/L at 20 \u00baC
Partition coefficient n-octanol/water (log value)	4.6
Vapour pressure	42 mm Hg ( 92 \u00baC)
Density and/or relative density	0.896g/mL at 25\u00baC(lit.)
Relative vapour density	4.76 (vs air)
Particle characteristics	no data available

### 10. Stability and reactivity

#### 10.1 Reactivity

no data available

#### 10.2 Chemical stability

On long exposure to air forms dangerous concentration of peroxide.

#### 10.3 Possibility of hazardous reactions

MODERATE, WHEN EXPOSED TO HEAT OR FLAME ...As a result of flow, agitation, etc., electrostatic charges can be generated. Saturated aliphatic hydrocarbons, such as DECAHYDRONAPHTHALENE, may be incompatible with strong oxidizing agents like nitric acid. Charring of the hydrocarbon may occur followed by ignition of unreacted hydrocarbon and other nearby combustibles. In other settings, aliphatic saturated hydrocarbons are mostly unreactive. They are not affected by aqueous solutions of acids, alkalis, most oxidizing agents, and most reducing agents. Oxidizes readily in air to form unstable peroxides that may explode spontaneously [Bretherick, 1979 p.151-154].

#### 10.4 Conditions to avoid

no data available

#### 10.5 Incompatible materials

... Can react with oxidizing materials.

#### 10.6 Hazardous decomposition products

When heated to decomposition it emits acrid smoke and fumes.

### 11. Toxicological information

#### Acute toxicity

Oral: LD50 Rat oral 4.2 g/kg

Inhalation: LC50 Rat inhalation 710 ppm/4 hr

Dermal: no data available

#### Skin corrosion/irritation

no data available

#### Serious eye damage/irritation

no data available

#### Respiratory or skin sensitization

no data available

#### Germ cell mutagenicity

no data available

#### Carcinogenicity

no data available

Reproductive toxicity  
 no data available  
 STOT-single exposure  
 no data available  
 STOT-repeated exposure  
 no data available  
 Aspiration hazard  
 no data available

## 12. Ecological information

### 12.1 Toxicity

Toxicity to fish: no data available  
 Toxicity to daphnia and other aquatic invertebrates: no data available  
 Toxicity to algae: no data available  
 Toxicity to microorganisms: no data available

### 12.2 Persistence and degradability

... Degradation in seawater by oil oxidizing micro-organisms: 13.6% breakdown after 21 days at 22°C in stoppered bottles containing a 1000 ppm mixture of alkanes, cycloalkanes, and aromatics

### 12.3 Bioaccumulative potential

An estimated BCF of 660 was calculated for decahydronaphthalene(SRC), using a water solubility of 0.889 mg/l(1) and a regression-derived equation(2). Over a test period of 8 weeks and using orange-red killifish (*Oryzias latipes*), BCF's of 839-2,380 at a test concn of 2.1 mg/l and 1,290-2,400 at a test concn of 0.21 mg/l were measured for the cis-isomer; BCF's of 1,170-3,050 at a test concn of 2.8 mg/l and 1,300-2,510 at a test concn of 0.28 mg/l were measured for the trans-isomer(3). According to a classification scheme(4), these BCF values suggest the potential for bioconcentration in aquatic organisms is very high(SRC).

### 12.4 Mobility in soil

The Koc of decahydronaphthalene is estimated as 4,600(SRC), using a water solubility of 0.889 mg/l(1) and a regression-derived equation(2). According to a classification scheme(3), this estimated Koc value suggests that decahydronaphthalene is expected to have slight mobility in soil.

### 12.5 Other adverse effects

no data available

## 13. Disposal considerations

### 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

#### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## 14. Transport information

### 14.1 UN Number

ADR/RID: UN1147

IMDG: UN1147

IATA: UN1147

### 14.2 UN Proper Shipping Name

ADR/RID: DECAHYDRONAPHTHALENE

IMDG: DECAHYDRONAPHTHALENE

IATA: DECAHYDRONAPHTHALENE

### 14.3 Transport hazard class(es)

ADR/RID: 3

IMDG: 3

IATA: 3

### 14.4 Packing group, if applicable

ADR/RID: III

IMDG: III

IATA: III

### 14.5 Environmental hazards

ADR/RID: yes

IMDG: yes

IATA: yes

### 14.6 Special precautions for user

no data available

### 14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

no data available

## 15. Regulatory information

### 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
decalin	decalin	91-17-8	none
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.

Vietnam National Chemical Inventory	Not Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)	Listed.

**Section 16: Other Information**

This safety data sheet should be used in conjunction with technical sheets. It does not replace them. The information given is based on our knowledge of this product, at the time of publication. It is given in good faith. The attention of the user is drawn to the possible risks incurred by using the product for any other purpose other than that for which it was intended. This does not in any way excuse the user from knowing and applying all the regulations governing his activity. It is the sole responsibility of the user to take all precautions required in handling the product. The aim of the mandatory regulations mentioned is to help the user to fulfill his obligations regarding the use of hazardous products.

